

# Preparation, Structural Analyses and Swelling Behavior of Amphiphilic Polymer Conetworks and Gels Based on Methacrylic Acid

Ph.D. Thesis

Gergely Áron Káli

Doctorate School of Chemistry  
Analytical Chemistry, Colloid- and Environmental Chemistry,  
Electrochemistry Program  
Eötvös Loránd Science University, Faculty of Natural Sciences  
Budapest, Hungary

Programme Leader  
Prof. Gyula Zsely

Supervisor  
Prof. Béla Iván

2009

## 1. Introduction and Aims

The investigation of the preparation, structure and properties of amphiphilic conetworks (APCN) as well as their possible applications have great interest in polymer science nowadays. These materials are comprised of covalently bonded, immiscible hydrophilic and hydrophobic polymer chains. Many unique and outstanding properties of APCNs, such as excellent mechanical strength, biocompatibility, swelling in polar and nonpolar solvents and the amorphous separated structures make them applicable either and in broader fields than homopolymer hydrogels. Another disadvantage: homopolymer poly(ethylene) hydrogels on the basis of the literature is that these materials undergo phase transition the gel collapse in the solution of different two or higher valence ions. As a consequence, there is a limitation of application hydrogels in high salt containing (for example some biological) systems.

A special class of APCNs is the group of model APCNs consisting of polymer chains with narrow molecular weight distribution and predetermined composition between the cross-links. The synthesis of model APCNs can be accomplished by the use of living polymerization techniques. Due to the well-known structure of the model APCNs, the structure-property relationships can be investigated better than in the case of randomly cross-linked APCNs. The first aim of this research was the preparation of three different model APCN series based on a well-known anionic poly(ethylene), the hydrophilic poly(methacrylic acid). In the first case, the hydrophobic segment was the commercially available, glassy poly(methyl methacrylate), while in the second case a non-commercial nitrory poly(2-hexyl-1-octyl methacrylate) was used. To the best of my knowledge, there is no any report in the literature on the synthesis and application of 2-hexyl-1-octyl methacrylate. The third hydrophobic monomer was a macromonomer: the nitroxy poly(isobutylene-methacrylate). The living polymerization method used as group transfer polymerization (GTP). The resulting conetwork series were widely investigated. My studies have been focused mainly on the investigation of the phase separated structures, then the mechanical and swelling properties as well as the phase separated structures. The swelling of these materials was also examined in solvents of biologically relevant salts.

Another aim of this work was the synthesis and swelling investigations of a known amphiphilic copolymer, poly(methacrylic acid)-*g*-poly(styrene) (PMAA-*g*-PS). The synthesis was carried out by the so-called macromonomer method and the swelling investigations were carried out in different salt solutions.

In the course of my research, I have also dealt with the problem related to with the hydrophilic monomer, methacrylic acid. In the two different synthesis strategies described above the same protected methacrylic acid cannot be used. Tetrahydrofurfuryl methacrylate (THFMA) is used in the GTP polymerization. This monomer is kinematically inelastic, so it is useless in the case of free radical polymerization in the macromonomer model. The trimethylsilyl methacrylate, used in the macromonomer method, is useless in GTP, because it can lead to the transfer of the trimethylsilyl group which is crucial in this polymerization process. Considering this problem, I aimed to prepare and use a new monomer, which is might be better suitable for obtaining methacrylic acid containing AFCNs.

## II. Applied Methods

The macromonomers, based on poly(styrene) were synthesized by quaternizing carboxonate, polymerization followed by several end group modification steps. The model copolymer was prepared by sequential monomer and cross-linker addition via Group Transfer Polymerization (GTP). Poly(methacrylic acid)-*g*-poly(styrene) AFCNs were synthesized by the macromonomer method via free radical copolymerization.

The linear polymers were analyzed by gel permeation chromatography (GPC) and <sup>1</sup>H NMR spectroscopy in terms of their molecular weight, polydispersity and composition. FTIR analyses were used to confirm the complete network formation and hydrolysis of the protective groups. The thermal behavior of the copolymer was investigated by thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC). The mechanical properties were determined by dynamic mechanical analysis (DMA). The mechanical separated structure of the copolymer was investigated by small-angle neutron scattering (SANS) and atomic force microscopy (AFM). Finally, the degree of swelling (R %) was determined arithmetically and was calculated as the solvent uptake divided by the mass of the dry network.

2

## III. New Scientific Results

1. Three series of model AFCNs were synthesized using sequential monomer and cross-linker addition via GTP. The hydrophilic monomer was methacrylic acid (MAA), while the three different hydrophobic monomers were methyl methacrylate (MMA), 2-hydroxy-1-octyl methacrylate (BOMA) and poly(styrene)-methacrylate (PBMA). The MAA units were introduced via the polymerization of tetrahydrofurfuryl methacrylate (THFMA) followed by the removal of the protecting tetrahydrofurfuryl group by acid hydrolysis after network formation. The linear copolymer precursors were analyzed by GPC and <sup>1</sup>H NMR spectroscopy in terms of their molecular weight and composition, both of which were found to be close to calculated values and the molecular weight distribution was narrow. FTIR spectroscopic analysis indicated complete polymerization of the EBDMA crosslinker vinyl units and complete hydrolysis of the THFMA units.

2. The degrees of swelling (DS) of the MMA, BOMA and PBMA based, MAA hydrophilic monomer containing copolymer networks were measured in water and in an organic solvent as a function of the degree of ionization (DI) of the MAA units. The results indicate that the DS of these copolymer networks increase by increasing the DI in water, and decrease by increasing the DI in THF. The effect of composition and architecture of the copolymer was studied as well, and it was found that these properties have influence on the DS. The DS of the fully ionized copolymer networks in water increase by increasing the MAA content. The effect of the architecture on the swelling behavior is more complex. The DS of the statistical copolymer based copolymer networks are always higher than that of block copolymer based model copolymer networks. The PK of the MAA units in the copolymer decreases by increasing the MAA content, but all the copolymer networks have higher PK values than the homopolymer PMAA network due to the effect of the hydrophobic environment in the copolymer. These results indicate that it is possible to influence the PK values of the copolymer by the hydrophobic content.

3

3. The phase mode AFM measurements of the dry MMA and BOMA based copolymers have proved the monophasic separated structure for the model copolymers. SAXS measurements have provided complementary information to AFM. The results of SAXS investigations indicated phase separated morphology in the case of block copolymer based model copolymers, where the phase sizes (distances between the scattering centers) are dependent on the composition. In the less ordered structures, such as statistical copolymers consisting of randomly cross-linked copolymer network, scattering maxima did not appear or only slight broad maxima occurred, which means that there is no phase separation in these materials.

4. The swelling behavior of the copolymers based on MMA and BOMA in the aqueous solution of NaCl, CaCl<sub>2</sub> and the mixture of these two salts were also investigated. The results show that these amphiphilic copolymers - in contrast to homopolymer poly(2-vinylpyridine) hydrogels - do not undergo discontinuous gel collapse even in very high salt concentrations, the change in the DS remains continuous. Thus, it can be concluded that these AFPCNs can be used in environments with very high salt concentrations.

5. A series of AFPCNs were successfully synthesized by the macroemulsion method in wide copolymer composition range using the bifunctional methacrylate-olefinic PIB macroemulsifier and MMA comonomer. The PIB macroemulsifier and the MMA comonomer have to combine solvent and the THPMA is thermally unstable to use in the case of free radical copolymerization. In this case, a protected MMA, the trimethylsilyl methacrylate was used, and the protecting groups were removed by acidic hydrolysis after copolymer formation. The elemental analyses of these poly(methacrylate acyl)-poly(styrene) (PMAA-*P*HB) copolymers after the deprotection confirmed that the preparation of this kind of copolymer is reproducible with a variety of compositions.

4

6. Another protected MMA, the ethoxyethyl methacrylate, described recently in the literature, have also been tried to produce a series of copolymers. Copolymers were successfully synthesized by free radical copolymerization of this comonomer and PIB-dimethacrylate. The deprotection was accomplished by acidic hydrolysis and thermolysis. FTIR investigations indicated better results for acyl hydrolysis. In the case of thermolysis of these copolymers, anhydride formation occurred. The temperatures of deprotection and decomposition were determined by TGA. The DSC graphs show the glass transition temperature of PIB, which result represents phase separation in these copolymers. The swelling behavior verifies the amphiphilic nature of the copolymer series. The AFPCNs swell in water as well as in hecane. Thus, it can be concluded ethoxyethyl methacrylate is a suitable monomer for copolymer synthesis by radical copolymerization with PIB-methacrylate.

7. The swelling behavior of PMAA-*P*HB copolymers was studied in aqueous solutions of Ca<sup>2+</sup>, Cu<sup>2+</sup> and La<sup>3+</sup> salts. In the case of Ca<sup>2+</sup> salt, the change in the DS remains continuous and lower than that of homopolymer hydrogels. With increasing CaCl<sub>2</sub> concentration, gel collapse does not occur. In the case of Cu<sup>2+</sup> and La<sup>3+</sup> salts, the change in the swelling degree is quite high in the range of small salt concentrations, but at higher level the degree of swelling will not change anymore. Thus it can be concluded that the investigated AFPCNs do not behave as it was expected on the basis of results with homopolymer poly(styrenes). These new results offer possibilities for these novel materials to be useful in applications in biological systems or for metal binding.

#### IV. Publications and Presentations

##### V.I. Publications

###### V.I.1. Publications related to this PhD work

1. G. Kell, B. Hahn, Swelling Response of Amphiphilic Copolymers to Salt Concentrations. *Proceedings of the Annual Symposium of Polymer Chemistry, Gdansk, Poland, 2006*
2. G. Kell, T. K. Georgiadis, H. Van, C.S. Parthenay, E. Lodon, V. Thomas, A.C.T. The Synthesis and Characterization of Novel Amphiphilic Model Copolymers based on Methacrylate Acid and

5

Methyl Methacrylate: Effects of Composition and Architecture

1. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, S. J. Park, M. Szwarc, Y. Thumana, J. C. Tiller, *Macromolecules*, **2007**, 40, 2192-2200.
2. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, Y. Thumana, J. C. Tiller, *Macromolecules*, **2007**, 40, 2192-2200.
3. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, E. Larson, Y. Thumana, J. C. Tiller, *Macromolecules*, **2007**, 40, 2192-2200.
4. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, E. Larson, Y. Thumana, J. C. Tiller, *Macromolecules*, **2007**, 40, 2192-2200.
5. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, E. Larson, Y. Thumana, J. C. Tiller, *Macromolecules*, **2007**, 40, 2192-2200.

### V.1.2 Publications not related to this PhD work

1. B. Vain, G. Kulk, G. Kulk, G. Halksvaak, Z. Jask, M. Szwarc, New functional  
*Polym. Mater. Sci. Eng.*, **2004**, 91, 971-98.
2. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
3. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
4. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
5. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.

### V.2 Publications

#### V.2.1 Publications related to this PhD work

1. G. Kulk, B. Vain, *Polymerization of polymer and polychlorinated benzoid azarides*,  
*PhD. Budapest*, 2005, April 26, (diss.)
2. G. Kulk, B. Vain, *Macromolecules*, **2007**, 40, 2192-2200.
3. G. Kulk, B. Vain, *Macromolecules*, **2007**, 40, 2192-2200.
4. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
5. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
6. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
7. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
8. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.

Macromolecules, 2007, 40, 2192-2200

9. Kulk, G., Vain, B., Geogagan, T., Parkinson, C., Larson, Y., Thumana, J., Tiller, C. *Macromolecules*, **2007**, 40, 2192-2200.
10. Kulk, G., Vain, B., Geogagan, T., Parkinson, C., Larson, Y., Thumana, J., Tiller, C. *Macromolecules*, **2007**, 40, 2192-2200.
11. Kulk, G., Vain, B., Geogagan, T., Parkinson, C., Larson, Y., Thumana, J., Tiller, C. *Macromolecules*, **2007**, 40, 2192-2200.
12. Kulk, G., Vain, B., Geogagan, T., Parkinson, C., Larson, Y., Thumana, J., Tiller, C. *Macromolecules*, **2007**, 40, 2192-2200.
13. Kulk, G., Vain, B., Geogagan, T., Parkinson, C., Larson, Y., Thumana, J., Tiller, C. *Macromolecules*, **2007**, 40, 2192-2200.

### V.1.2 Publications not related to this PhD work

14. B. Vain, G. Kulk, G. Kulk, G. Halksvaak, Z. Jask, M. Szwarc, New functional  
*Polym. Mater. Sci. Eng.*, **2004**, 91, 971-98.
15. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
16. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
17. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.
18. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, M. Szwarc, *Macromol. Rapid Commun.*, **2007**, 28, 1381-1393.

### V.2 Publications

#### V.2.2 Publications related to this PhD work

19. G. Kulk, B. Vain, *Polymerization of polymer and polychlorinated benzoid azarides*,  
*PhD. Budapest*, 2005, April 26, (diss.)
20. G. Kulk, B. Vain, *Macromolecules*, **2007**, 40, 2192-2200.
21. G. Kulk, B. Vain, *Macromolecules*, **2007**, 40, 2192-2200.
22. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
23. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
24. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.
25. G. Kulk, T. K. Geogagan, B. Vain, C. S. Parkinson, *Macromolecules*, **2007**, 40, 2192-2200.

26. G. Kati, M. Szecsenyi, A. Balazs, H. Imai, Synthesis of New, Branched Polyimides, Syntheses by Quaternary Carbonate Polymerization, IUPAC International Symposium on Issue: Chemistry of Cellulose, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
27. C. Kádor, G. Kati, P. W. Goik, M. Huzar, J. J. Rodriguez, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
28. Coordination Soluble, Of Amphiphilic Copolyols Based On N-vinylcarbazole, 11th Booklet of IACS, Budapest, Hungary, 2006
29. H. Imai, G. Erdödi, P. W. Goik, G. Kati, Gy. Kozsa, A. Salgó, I. Szanka, M. Szecsenyi, New method for new multifunctional hyperbranched polymers, *Adv. Materials Symposium on Chemistry of Cellulose*, 2007, (in press)
30. H. Imai, G. Erdödi, P. W. Goik, M. Huzar, B. J. M. H. J. J. Rodriguez, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
31. H. Imai, G. Erdödi, P. W. Goik, M. Huzar, J. J. Rodriguez, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
32. S. Szabó, I. Szanka, Gy. Szarka, K. Veszteg, [3] amine-terminated polymer end-caps, *and 6g monobranched amine-phenylene, A Magyar Tudomány* 2007, Budapest 2007, november 12.
33. G. Erdödi, G. Kati, P. W. Goik, M. Huzar, A. Hölter, H. Imai, G. Kati, P. Marcy, V. Palkó, A. monobranched amine, I.ETI Innovációs Nap, Budapest, 2008, téma: 6g (in press)
34. C. Kádor, G. Kati, A. Dórnai, K. Z. Péterfy, H. Imai, N-vinylcarbazol alapú and 6g monobranched amine-phenylene, *ETI Innovációs Nap*, Budapest, 2008, téma: 6g (in press)
35. S. Szabó, Gy. Kozsa, A. Salgó, P. W. Goik, M. Huzar, Gy. Kati, I. Szanka, M. Szecsenyi, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
36. S. Szabó, Gy. Kozsa, A. Salgó, P. W. Goik, Gy. Kati, I. Szanka, M. Szecsenyi, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
37. S. Szabó, Gy. Kozsa, A. Salgó, P. W. Goik, Gy. Kati, I. Szanka, M. Szecsenyi, *Journal of Polymer Science: Part A: Polymer Chemistry* 2007, (in press)
38. Gy. Kozsa, G. Kati, M. Szecsenyi, H. Imai, The effect of reaction conditions on the formation of hyperbranched polymers by quaternary carbonate polymerization, *Kolloidok*, 13(1), november 2006, (in press)